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Classical Thermodynamics of Non-Electrolyte Solutions Problems And Solutions On Thermodynamics And Statistical Mechanics (the Volume Is
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Mathematical Thermodynamics of Complex Fluids Thermodynamics in Bioenergetics

The Solutions Manual to Accompany Elements of Physical Chemistry 7th edition contains full worked solutions to all end-of-chapter discussion questions and exercises featured in the book. The simulation and optimization of processes assumes that the thermodynamic properties and phase equilibria of the mixtures concerned are well known. This knowledge is still based upon experimentation, but it is also the result of calculation methods based on the principles of thermodynamics that govern them, insure their coherence, and confer upon them a wide range of application. This text is concerned primarily with the description of these methods and their evolution. It devotes extensive space to fundamental concepts and places particular emphasis on the "models" that, although based on simplified concepts of the subject matter at the molecular level, have "predictive" character. Computational examples are used to explain the application of these concepts and models. Contents: 1. Principles. Thermodynamic functions. The ideal gas. 2. Properties of pure substances. 3. Predicting thermodynamic properties of pure substances. General principles. Corresponding states. Group contributions. 4. Equations of state. 5. Characterization of mixtures. 6. Mixtures: liquid-vapor equilibria. 7. Deviations from ideality in the liquid phase.

8. Application of equations of state to mixtures. Calculation of liquid-vapor equilibria under pressure. 9. Liquid-liquid and liquid-liquid-vapor equilibria. 10. Fluid-solid equilibria. Crystallization. Hydrates. 11. Polymer solutions and alloys. 12. Multicomponent mixtures. 13. Chemical reactions. Appendixes. Index. Bibliography. As one of the results of an ambitious project, this handbook provides a well-structured directory of globally available software tools in the area of Integrated Computational Materials Engineering (ICME). The compilation covers models, software tools, and numerical methods allowing describing electronic, atomistic, and mesoscopic phenomena, which in their combination determine the microstructure and the properties of materials. It reaches out to simulations of component manufacture comprising primary shaping, forming, joining, coating, heat treatment, and machining processes. Models and tools addressing the in-service behavior like fatigue, corrosion, and eventually recycling complete the compilation. An introductory overview is provided for each of these different modelling areas highlighting the relevant phenomena and also discussing the current state for the different simulation approaches. A must-have for researchers, application engineers, and simulation software providers seeking a holistic overview about the current state of the art in a huge variety of modelling topics. This handbook equally serves as a reference manual for academic and commercial software developers and providers, for industrial users of simulation software, and for decision makers seeking to optimize their production by simulations. In view of its sound introductions into the different fields of materials physics, materials chemistry, materials engineering and materials processing it also serves as a tutorial for students in the emerging discipline of ICME, which requires a broad view on things and at least a basic education in adjacent fields. This book considers molecular structural information, statistical methods and thermodynamic measurements, and the ways in which the relative role of each differs from another. By putting together selected papers in a single publication, the book highlights the cohesive aspects of certain advances through time and development, and can aid historical studies. Several papers from journals not widely circulated can also be found in this selection of papers. This book consists of a number of papers regarding the thermodynamics and structure of multicomponent systems that we have published during the last decade. Even though they involve different topics and different systems, they have something in common which can be considered as the “signature” of the present book. First, these papers are concerned with “difficult” or very nonideal systems, i. e. systems with very strong interactions (e. g. , hyd- gen bonding) between components or systems with large differences in the partial molar volumes of the components (e. g. , the aqueous solutions of proteins), or systems that are far from “normal” conditions (e. g. , critical or near-critical mixtures). Second, the conventional thermodynamic methods are not sufficient for the accurate treatment of these mixtures. Last but not least, these systems are of interest for the pharmaceutical, biomedical, and related industries. In order to meet the thermodynamic challenges involved in these complex mixtures, we employed a variety of traditional methods but also new methods, such as the fluctuation theory of Kirkwood and Buff and ab initio quantum mechanical techniques. The Kirkwood-Buff (KB) theory is a rigorous formalism which is free of any of the approximations usually used in the thermodynamic treatment of multicomponent systems. This theory appears to be very fruitful when applied to the above mentioned “difficult” systems. Thermodynamics Seventh Edition covers the basic principles of thermodynamics while presenting a wealth of real-world engineering examples so students get a feel for how thermodynamics is applied in engineering practice. This text helps students develop an intuitive understanding of thermodynamics by emphasizing the physics and physical arguments. Cengel/Boles explore the various facets of thermodynamics through careful explanations of concepts and its use of numerous practical examples and figures, having students develop necessary skills to bridge the gap between knowledge and the confidence to properly apply knowledge. The media package for this text is extensive, giving users a large variety of supplemental resources to choose from. A Student Resources DVD is packaged with each new copy of the text and contains the popular Engineering Equation Solver (EES) software. McGraw-Hill's new Connect is available to students and instructors.

Connect is a powerful, web-based assignment management system that makes creating and grading assignments easy for instructors and learning convenient for students. It saves time and makes learning for students accessible anytime, anywhere. With Connect, instructors can easily manage assignments, grading, progress, and students receive instant feedback from assignments and practice problems. Practical Chemical Thermodynamics for Geoscientists covers classical chemical thermodynamics and focuses on applications to practical problems in the geosciences, environmental sciences, and planetary sciences. This book will provide a strong theoretical foundation for students, while also proving beneficial for earth and planetary scientists seeking a review of thermodynamic principles and their application to a specific problem. Strong theoretical foundation and emphasis on applications Numerous worked examples in each chapter Brief historical summaries and biographies of key thermodynamicists-including their fundamental research and discoveries Extensive references to relevant literature The Solutions Manual to Accompany Elements of Physical Chemistry 7th edition contains full worked solutions to all end-of-chapter discussion questions and exercises featured in the book. The manual provides helpful comments and friendly advice to aid understanding. It is also a valuable resource for any lecturer who wishes to use the extensive selection of exercises featured in the text to support either formative or summative assessment, and wants labour-saving, ready access to the full solutions to these questions. Classical Thermodynamics of Non-Electrolyte Solutions covers the historical development of classical thermodynamics that concerns the properties of vapor and liquid solutions of non-electrolytes. Classical thermodynamics is a network of equations, developed through the formal logic of mathematics from a very few fundamental postulates and leading to a great variety of useful deductions. This book is composed of seven chapters and begins with discussions on the fundamentals of thermodynamics and the thermodynamic properties of fluids. The succeeding chapter presents the equations of state for the calculation of the thermodynamic behavior of constant-composition fluids, both liquid and gaseous. These topics are followed by surveys of the mixing of pure materials to form a solution under conditions of constant temperature and pressure. The discussion then shifts to general equations for calculation of partial molal properties of homogeneous binary systems. The last chapter considers the approach to equilibrium of systems within which composition changes are brought about either by mass transfer between phases or by chemical reaction within a phase, or by both. Now in its seventh edition, Fundamentals of Thermodynamics continues to offer a comprehensive and rigorous treatment of classical thermodynamics, while retaining an engineering perspective. With concise, applications-oriented discussion of topics and self-test problems the text encourages students to monitor their own comprehension. The seventh edition is updated with additional examples, homework problems, and illustrations to increase student understanding. The text lays the groundwork for subsequent studies in fields such as fluid mechanics, heat transfer and statistical thermodynamics, and prepares students to effectively apply thermodynamics in the practice of engineering. Based on a university course, this book provides an exposition of a large spectrum of geological, geochemical and geophysical problems that are amenable to thermodynamic analysis. It also includes selected problems in planetary sciences, relationships between thermodynamics and microscopic properties, particle size effects, methods of approximation of thermodynamic properties of minerals, and some kinetic ramifications of entropy production. The textbook will enable graduate students and researchers alike to develop an appreciation of the fundamental principles of thermodynamics, and their wide ranging applications to natural processes and systems. Created for engineers and students working with pure polymers and polymer solutions, this handbook provides up-to-date, easy to use methods to obtain specific volumes and phase equilibrium data. A comprehensive database for the phase equilibria of a wide range of polymer-solvent systems, and PVT behavior of pure polymers are given, as are accurate predictive techniques using group contributions and readily available pure component data. Two computer programs on diskettes are included. POLYPROG implements procedures given for prediction and correlation for specific volume of pure polymer liquids and calculation of vapor-liquid equilibria (VLE) of polymer solutions.

POLYDATA provides an easy method of accessing the data contained in the many databases in the book. Both disks require a computer with a math coprocessor. This handbook is a valuable resource in the design and operation of many polymer processes, such as polymerization, devolatilization, drying, extrusion, and heat exchange. Special Details: Hardcover with Disks. Special offer: Purchase this book along with X-131, Handbook of Diffusion and Thermal Properties of Polymers and Polymer Solutions and receive a 20 percent discount off the list or member price. Outstanding text focuses on physical technique of thermodynamics, typical problems, and significance and use of thermodynamic potential. Mathematical apparatus, first law of thermodynamics, second law and entropy, more. 1965 edition. This text is the published version of many of the talks presented at two symposiums held as part of the Southeast Regional Meeting of the American Chemical Society (SERMACS) in Knoxville, TN in October, 1999. The Symposiums, entitled Solution Thermodynamics of Polymers and Computational Polymer Science and Nanotechnology, provided outlets to present and discuss problems of current interest to polymer scientists. It was, thus, decided to publish both proceedings in a single volume. The first part of this collection contains printed versions of six of the ten talks presented at the Symposium on Solution Thermodynamics of Polymers organized by Yuri B. Melnichenko and W. Alexander Van Hook. The two sessions, further described below, stimulated interesting and provocative discussions. Although not every author chose to contribute to the proceedings volume, the papers that are included faithfully represent the scope and quality of the symposium. The remaining two sections are based on the symposium on Computational Polymer Science and Nanotechnology organized by Mark D. Dadmun, Bobby G. Sumpter, and Don W. Noid. A diverse and distinguished group of polymer and materials scientists, biochemists, chemists and physicists met to discuss recent research in the broad field of computational polymer science and nanotechnology. The two-day oral session was also complemented by a number of poster presentations. The first article of this section is on the important subject of polymer blends. M. D. The classic guide to mixtures, completely updated with new models, theories, examples, and data. Efficient separation operations and many other chemical processes depend upon a thorough understanding of the properties of gaseous and liquid mixtures. Molecular Thermodynamics of Fluid-Phase Equilibria, Third Edition is a systematic, practical guide to interpreting, correlating, and predicting thermodynamic properties used in mixture-related phase-equilibrium calculations. Completely updated, this edition reflects the growing maturity of techniques grounded in applied statistical thermodynamics and molecular simulation, while relying on classical thermodynamics, molecular physics, and physical chemistry wherever these fields offer superior solutions. Detailed new coverage includes: Techniques for improving separation processes and making them more environmentally friendly. Theoretical concepts enabling the description and interpretation of solution properties. New models, notably the lattice-fluid and statistical associated-fluid theories. Polymer solutions, including gas-polymer equilibria, polymer blends, membranes, and gels. Electrolyte solutions, including semi-empirical models for solutions containing salts or volatile electrolytes. Coverage also includes: fundamentals of classical thermodynamics of phase equilibria; thermodynamic properties from volumetric data; intermolecular forces; fugacities in gas and liquid mixtures; solubilities of gases and solids in liquids; high-pressure phase equilibria; virial coefficients for quantum gases; and much more. Throughout, Molecular Thermodynamics of Fluid-Phase Equilibria strikes a perfect balance between empirical techniques and theory, and is replete with useful examples and experimental data. More than ever, it is the essential resource for engineers, chemists, and other professionals working with mixtures and related processes. The main goal of this book is to provide an overview of the state of the art in the mathematical modeling of complex fluids, with particular emphasis on its thermodynamical aspects. The central topics of the text, the modeling, analysis and numerical simulation of complex fluids, are of great interest and importance both for the understanding of various aspects of fluid dynamics and for its applications to special real-world problems. New emerging trends in the subject are highlighted with the intent to inspire and motivate young researchers and PhD students. Problems in

Metallurgical Thermodynamics and Kinetics provides an illustration of the calculations encountered in the study of metallurgical thermodynamics and kinetics, focusing on theoretical concepts and practical applications. The chapters of this book provide comprehensive account of the theories, including basic and applied numerical examples with solutions. Unsolved numerical examples drawn from a wide range of metallurgical processes are also provided at the end of each chapter. The topics discussed include the three laws of thermodynamics; Clausius-Clapeyron equation; fugacity, activity, and equilibrium constant; thermodynamics of electrochemical cells; and kinetics. This book is beneficial to undergraduate and postgraduate students in universities, polytechnics, and technical colleges. As the title suggests, we introduce a novel differential approach to solution thermodynamics and use it for the study of aqueous solutions. We evaluate the quantities of higher order derivative than the normal thermodynamic functions. We allow these higher derivative data speak for themselves without resorting to any model system. We thus elucidate the molecular processes in solution, (referred to in this book “mixing scheme”), to the depth equal to, if not deeper, than that gained by spectroscopic and other methods. We show that there are three composition regions in aqueous solutions of non-electrolytes, each of which has a qualitatively distinct mixing scheme. The boundary between the adjacent regions is associated with an anomaly in the third derivatives of G . The loci of the anomalies in the temperature-composition field form the line sometimes referred as “Koga line”. We then take advantage of the anomaly of a third derivative quantity of 1-propanol in the ternary aqueous solution, 1-propanol – sample species – H_2O . We use its induced change as a probe of the effect of a sample species on H_2O . In this way, we clarified what a hydrophobe, or a hydrophile, and in turn, an amphiphile, does to H_2O . We also apply the same methodology to ions that have been ranked by the Hofmeister series. We show that the kosmotropes (salting out, or stabilizing agents) are either hydrophobes or hydration centres, and that chaotropes (salting in, or destabilizing agents) are hydrophiles. A new differential approach to solution thermodynamics

A particularly clear elucidation of the mixing schemes in aqueous solutions
A clear understandings on the effects of hydrophobes, hydrophiles, and amphiphiles to H_2O
A clear understandings on the effects of ions on H_2O in relation to the Hofmeister effect
A new differential approach to studies in multi-component aqueous solutions
Complexity and Complex Chemo-Electric Systems presents an analysis and synthesis of chemo-electric systems, providing insights on transports in electrolytes, electrode reactions, electrocatalysis, electrochemical membranes, and various aspects of heterogeneous systems and electrochemical engineering. The book describes the properties of complexity and complex chemo-electric systems as the consequence of formulations, definitions, tools, solutions and results that are often consistent with the best performance of the system. The book handles cybernetics, systems theory and advanced contemporary techniques such as optimal control, neural networks and stochastic optimizations (adaptive random search, genetic algorithms, and simulated annealing). A brief part of the book is devoted to issues such as various definitions of complexity, hierarchical structures, self-organization examples, special references, and historical issues. This resource complements Sieniutycz’ recently published book, Complexity and Complex Thermodynamic Systems, with its inclusion of complex chemo-electric systems in which complexities, emergent properties and self-organization play essential roles. Covers the theory and applications of complex chemo-electric systems through modeling, analysis, synthesis and optimization Provides a clear presentation of the applications of transport theory to electrolyte solutions, heterogeneous electrochemical systems, membranes, electro-kinetic phenomena and interface processes Includes numerous explanatory graphs and drawings that illustrate the properties and complexities in complex chemo-electric systems Written by an experienced expert in the field of advanced methods in thermodynamics and related aspects of macroscopic physics This book is based on a set of notes developed over many years for an introductory course taught to seniors and entering graduate students in materials science. An Introduction to Aspects of Thermodynamics and Kinetics Relevant to Materials Science is about the application of thermodynamics and kinetics to solve problems within Materials Science. Emphasis is to provide a physical understanding of the

phenomenon under discussion, with the mathematics presented as a guide. The problems are used to provide practice in quantitative application of principles, and also to give examples of applications of the general subject matter to problems having current interest and to emphasize the important physical concepts. End of chapter problems are included, as are references, and bibliography to reinforce the text. This book provides students with the theory and mathematics to understand the important physical understanding of phenomena. Based on a set of notes developed over many years for an introductory course taught to seniors and entering graduate students in materials science Provides students with the theory and mathematics to understand the important physical understanding of phenomena Includes end of chapter problems, references, and bibliography to reinforce the text This book was planned and written with one central goal in mind: to demonstrate that statistical thermodynamics can be used successfully by a broad group of scientists, ranging from chemists through biochemists to biologists, who are not and do not intend to become specialists in statistical thermodynamics. The book is addressed mainly to graduate students and research scientists interested in designing experiments the results of which may be interpreted at the molecular level, or in interpreting such experimental results. It is not addressed to those who intend to practice statistical thermodynamics per se. With this goal in mind, I have expended a great deal of effort to make the book clear, readable, and, I hope, enjoyable. This does not necessarily mean that the book as a whole is easy to read. The first four chapters are very detailed. The last four become progressively more difficult to read, for several reasons. First, presuming that the reader has already acquired familiarity with the methods and arguments presented in the first part, I felt that similar arguments could be skipped later on, leaving the details to be filled in by the reader. Second, the systems themselves become progressively more complicated as we proceed toward the last chapter. Chemical Thermodynamics: Principles and Applications presents a thorough development of the principles of thermodynamics--an old science to which the authors include the most modern applications, along with those of importance in developing the science and those of historical interest. The text is written in an informal but rigorous style, including anecdotes about some of the great thermodynamicists (with some of whom the authors have had a personal relationship), and focuses on "real" systems in the discussion and figures, in contrast to the generic examples that are often used in other textbooks. The book provides a basic review of thermodynamic principles, equations, and applications of broad interest. It covers the development of thermodynamics as one of the pre-eminent examples of an exact science. A discussion of the standard state that emphasizes its significance and usefulness is also included, as well as a more rigorous and indepth treatment of thermodynamics and discussions of a wider variety of applications than are found in more broadly based physical chemistry undergraduate textbooks. Combined with its companion book, Chemical Thermodynamics: Advanced Applications, the practicing scientist will have a complete reference set detailing chemical thermodynamics. Outlines the development of the principles of thermodynamics, including the most modern applications along with those of importance in developing the science and those of historical interest Provides a basic review of thermodynamic principles, equations, and applications of broad interest Treats thermodynamics as one of the preeminent examples of an exact science Provides a more rigorous and indepth treatment of thermodynamics and discussion of a wider variety of applications than are found in more broadly based physical chemistry undergraduate textbooks Includes examples in the text and exercises and problems at the end of each chapter to assist the student in learning the subject Provides a complete set of references to all sources of data and to supplementary reading sources Basic research and new manufacturing methods have led to high nitrogen steels (HNS), a promising new group of materials for use in advanced applications in mechanical and chemical engineering. The book deals with the atomic structure, constitution, properties, manufacturing and application of martensitic, austenitic, duplex and dualphase steels of superior strength and corrosion resistance. Combining metallurgy and engineering aspects. It gives a detailed overview and presents new results on HNS. The book is intended for scientists as well as technologists, who will find stimulating information. Thermodynamics in

Bioenergetics aims to supply students with the knowledge and understanding of the critical concepts and theories that are needed in the biochemistry and bioenergetics fields. Biochemical reactions highlighting thermodynamics, chemical kinetics, and enzymes are addressed in the text. Author, Jean-Louis Burgot, guides the reader through the starting points, strategy description, and theory results to facilitate their comprehension of the theories and examples being discussed in the book. Also discussed in the text are the notions of Gibbs energy, entropy, and exergonic and endergonic reactions.

Volume 76 of *Reviews in Mineralogy and Geochemistry* presents an extended review of the topics conveyed in a short course on Geothermal Fluid Thermodynamics held prior to the 23rd Annual V.M. Goldschmidt Conference in Florence, Italy (August 24-25, 2013). It covers Thermodynamics of Geothermal Fluids, The Molecular-Scale Fundament of Geothermal Fluid Thermodynamics, Thermodynamics of Aqueous Species at High Temperatures and Pressures: Equations of State and Transport Theory, Mineral Solubility and Aqueous Speciation Under Hydrothermal Conditions to 300 °C – The Carbonate System as an Example, Thermodynamic Modeling of Fluid-Rock Interaction at Mid-Crustal to Upper-Mantle Conditions, Speciation and Transport of Metals and Metalloids in Geological Vapors, Solution Calorimetry Under Hydrothermal Conditions, Structure and Thermodynamics of Subduction Zone Fluids from Spectroscopic Studies and Thermodynamics of Organic Transformations in Hydrothermal Fluids.

The material for these volumes has been selected from the past twenty years' examination questions for graduate students at University of California at Berkeley, Columbia University, the University of Chicago, MIT, State University of New York at Buffalo, Princeton University and University of Wisconsin. Because classical thermodynamics evolved into many branches of science and engineering, most undergraduate courses on the subject are taught from the perspective of each area of specialization. General Thermodynamics combines elements from mechanical and chemical engineering, chemistry (including electrochemistry), materials science, and biology to present a unique and thorough treatment of thermodynamics that is broader in scope than other fundamental texts. This book contains classroom-tested materials designed to meet the academic requirements for students from a variety of scientific and engineering backgrounds in a single course. The first half focuses on classical concepts of thermodynamics, whereas the latter half explores field-specific applications, including a unique chapter on biothermodynamics. The book's methodology is unified, concise, and multidisciplinary, allowing students to understand how the principles of thermodynamics apply to all technical fields that touch upon this most fundamental of scientific theories. It also offers a rigorous approach to the quantitative aspects of thermodynamics, accompanied by clear explanations to help students transition smoothly from the physical concepts to their mathematical representations. Each chapter contains numerous worked examples taken from different engineering applications, illustrations, and an extensive set of exercises to support the material. A complete solutions manual is available to professors with qualifying course adoptions. Thermodynamics deals with energy levels and the transfer of energy between states of matter, and is therefore fundamental to all branches of science. This edition provides a relatively advanced treatment of the subject, specifically tailored for the interests of the Earth sciences. The first four chapters explain all necessary concepts, using a simple graphical approach. Throughout the rest of the book the author emphasizes the use of thermodynamics to construct mathematical simulations of real systems. This helps to make the many abstract concepts acceptable. Many computer programs are mentioned and used throughout the text, especially SUPCRT92, a widely used source of thermodynamic data. An associated website includes links to useful information sites and computer programs and problem sets. Building on the more elementary material in the first edition, this textbook will be ideal for advanced undergraduate and graduate students in geology, geochemistry, geophysics and environmental science. Accompanying DVD-ROM contains the Limited Academic Version of EES (Engineering Equation Solver) software with scripted solutions to selected text problems. A comprehensive introduction, examining both macroscopic and microscopic aspects of the subject, the book applies the theory of thermodynamics to a broad range of materials; from metals, ceramics and other inorganic materials to geological

materials. Focusing on materials rather than the underlying mathematical concepts of the subject, this book will be ideal for the non-specialist requiring an introduction to the energetics and stability of materials. Macroscopic thermodynamic properties are linked to the underlying microscopic nature of the materials and trends in important properties are discussed. A unique approach covering both macroscopic and microscopic aspects of the subject. Authors have worldwide reputations in this area. Fills a gap in the market by featuring a wide range of real up-to-date examples and covering a large amount of materials. This book is an excellent companion to *Chemical Thermodynamics: Principles and Applications*. Together they make a complete reference set for the practicing scientist. This volume extends the range of topics and applications to ones that are not usually covered in a beginning thermodynamics text. In a sense, the book covers a "middle ground" between the basic principles developed in a beginning thermodynamics textbook, and the very specialized applications that are a part of an ongoing research project. As such, it could prove invaluable to the practicing scientist who needs to apply thermodynamic relationships to aid in the understanding of the chemical process under consideration. The writing style in this volume remains informal, but more technical than in *Principles and Applications*. It starts with Chapter 11, which summarizes the thermodynamic relationships developed in this earlier volume. For those who want or need more detail, references are given to the sections in *Principles and Applications* where one could go to learn more about the development, limitations, and conditions where these equations apply. This is the only place where *Advanced Applications* ties back to the previous volume. Chapter 11 can serve as a review of the fundamental thermodynamic equations that are necessary for the more sophisticated applications described in the remainder of this book. This may be all that is necessary for the practicing scientist who has been away from the field for some time and needs some review. The remainder of this book applies thermodynamics to the description of a variety of problems. The topics covered are those that are probably of the most fundamental and broadest interest. Throughout the book, examples of "real" systems are used as much as possible. This is in contrast to many books where "generic" examples are used almost exclusively. A complete set of references to all sources of data and to supplementary reading sources is included. Problems are given at the end of each chapter. This makes the book ideally suited for use as a textbook in an advanced topics course in chemical thermodynamics. An excellent review of thermodynamic principles and mathematical relationships along with references to the relevant sections in *Principles and Applications* where these equations are developed. Applications of thermodynamics in a wide variety of chemical processes, including phase equilibria, chemical equilibrium, properties of mixtures, and surface chemistry. Case-study approach to demonstrate the application of thermodynamics to biochemical, geochemical, and industrial processes. Applications at the "cutting edge" of thermodynamics. Examples and problems to assist in learning. Includes a complete set of references to all literature sources.

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